

Development of a tool for the complete first-principles prediction of finite-temperature and generalized initial-state UV absorption spectra: Application to NO₂







Richard Dawes, Ernesto Quintas-Sánchez, Steve Ndengué

NO₂ has a notoriously complex UV absorption spectrum



- First explore consistency of available data
- Experiments are challenging due to photodissociation to NO + O beginning at ~25000 cm⁻¹ and equilibrium with N₂O₄.
- Mainz database has several spectra recorded by different people, temperatures, resolution etc. <u>http://satellite.mpic.de/spectral_atlas/cross_sections/Nitrogen%20oxides/NO2.spc</u>
- Data recorded by Bogumil in 2003 includes the lowest temperature available (203 K).



NO₂ has a notoriously complex UV absorption spectrum



 Several datasets are available near 220 K and are fairly consistent except for differences in resolution

NO₂ has a notoriously complex UV absorption spectrum



Several datasets are available around 294 K and are more consistent, with that by Vandaele (1998) having the highest resolution (2 cm⁻¹). A.C. Vandaele, C. Hermans, P.C. Simon, M. Carleer, R. Colins, S. Fally, M.F. Mérienne, A. Jenouvrier, and B. Coquart, "Measurements of the NO₂ absorption cross-sections from 42000 cm⁻¹ to 10000 cm⁻¹ (238-1000 nm) at 220 K and 294 K", J. Quant. Spectrosc. Radiat. Transfer 59, 171-184 (1998); DOI: 10.1016/S0022-4073(97)00168-4

Electronic Schrödinger Equation: Single reference method hierarchy



Chem. Soc. Rev. 2012, 41, 6259

MISSOUR

Molecular nitrogen: N₂



What happens if we don't account for changing of important electron configurations in N_2 as it dissociates?



Single-reference breakdown



Divergence of CCSD(T) occurs for $r_{NN} > 2$ Å. X. Li and J. Paldus, *J. Chem. Phys.* **129**, 054104 (2008).



Multireference Calculations

- Multi-reference methods (e.g. MRCI) are usually required when forming/breaking bonds, or even at particular minima when the electronic structure is not welldescribed by a single configuration (e.g. Ozone)
- Most convenient way to describe multiple states of the same symmetry and spin
- Multireference explicitly correlated F12 methods are available (CASPT2-F12, MRCI-F12).
- No common cost-effective strategies for higher-order correlation
 - Davidson correction to approximately restore size-consistency and extensivity is sometimes said to account for higher-order correlation, but often is not as accurate for thermochemistry as CCSD(T)
- Internal contraction used in Molpro to greatly reduce cost has been shown to affect the attractiveness of some PESs in dynamically important ways (e.g. HO₂, Harding, Klippenstein, Lischka, Shepard-*Theo. Chem. Acc.* **133** 1 (2014), O₃, A. Powell, N.S. Dattani, RFK Spada, FBC Machado, H. Lischka, R. Dawes, J. Chem. Phys. **147**, 094306 (2017).)



Electronic Schrödinger Equation: Multireference method hierarchy



Adapted from Chem. Soc. Rev. 2012, 41, 6259



Dynamically-weighted state-averaged multireference electronic structure theory

$$\overline{E}^{SA-MCSCF} = \sum_{i=0}^{n} \frac{W_i}{W} E_i^{MCSCF} \qquad W = \sum_{i=0}^{n} W_i$$

Dynamic weighting¹

$$w_i = \cosh^2(\Delta E_{i0} / \beta)$$
 $\Delta E_{i0} = E_i - E_0$

Generalized dynamic weighting²

$$w_i = \cosh^2 \left(\Delta E_{ij} / \beta \right) \qquad \Delta E_{ij} = |E_i - E_j|$$



¹M.P. Deskevich, D.J. Nesbitt and H-J. Werner, JCP **120** 7281 (2004) ²R. Dawes, A.W. Jasper...S.A. Reid *et al.* J. Phys. Chem. Lett. **1** 641 (2010)



Importance of Dynamic Weighting: BeOBe → BeO + Be

1 state

23 states equal weights



In single state calc (at left), avoided crossing with higher lying state (blue oval, at right) causes discontinuity.

In fixed weights calc (at right) crossing with higher lying state (red oval) causes discontinuity cascade affecting ground state.

Six lowest A_1 states (co-linear asymmetric stretch C_{2V})



Importance of Dynamic Weighting: BeOBe → BeO + Be

23 states dynamic weights

-103.85 -103.85 ·********** -103.90 -103.90 -103.95 -103.95 -104.00 -104.00 Energy (a.u.) Energy (a.u.) -104.05 -104.05 -104.10 -104.10 -104.15 -104.15 -104.20 -104.20 -104.25 -104.25 -104.30 -104.30 -104.35 -104.35 1.00 1.50 2.00 2.50 3.00 1.00 1.50 2.00 2.50 3.00 3.50 3.50 rBeO (Angstroms) rBeO (Angstroms)

23 states equal weights

With dynamic weights (at right), crossing still occurs (blue oval), but due to negligible weight, no disruption results.

R.Dawes and S. Ndengué, "Single- and multireference electronic structure calculations for constructing potential energy surfaces", *International Reviews in Physical Chemistry* **35**, 441 (2016).



Dynamic weighting applied to 18 doublet states of NO₂

Each plot shows a different cut through the PESs Calculations include 8 A' and 10 A'' states Plot only show 8 A' (open symbols) and lowest 3 A'' states (red)



-states form two blocks at NO + O limit

Bend: bond distances held at equilibrium -X and A states cross around 107° -X and B states form RT pair at 180° 2 A'' states cross in vicinity of FC region

Symmetric stretch: angle held at equilibrium, 133° -all 18 included states are degenerate at separate atoms asymptote (ground state atoms) as expected



Normal mode displaced geometries for MRCI-F12(Q)/VTZ-F12 method





Note – phase of bend is such that positive displacement *contracts* the angle



Effect of dynamic correlation on state gaps and crossing points



Filled symbols – A' states Open symbols – A'' states Vertical gaps CASSCF level 26588, 26375, 28747 Vertical gaps MRCI level 26694, 24104, 28883 Vertical gaps MRCI(Q) level 26593, 23283, 29145



Effect of dynamic correlation on state gaps and crossing points



Energies along asymmetric stretch coordinate are symmetric with respect to positive/negative displacement. The two most relevant A'' states don't cross along this coordinate. The A'' energies and spacing are sensitive to dynamic electron correlation



Dipoles and Diabatization



$$\mathbf{U} = \begin{pmatrix} \cos\theta & \sin\theta \\ -\sin\theta & \cos\theta \end{pmatrix}$$

 $\theta = \frac{1}{4} \tan^{-1} \left(\frac{G}{-F} \right),$ $F = |\boldsymbol{\mu}_{12}^0|^2 - |\boldsymbol{\mu}_{11}^0 - \boldsymbol{\mu}_{22}^0|^2 / 4,$ $G = \boldsymbol{\mu}_{12}^0 \cdot (\boldsymbol{\mu}_{11}^0 - \boldsymbol{\mu}_{22}^0),$

 $\boldsymbol{\mu}_{JK}^{0} = \langle \Phi_{J} | \hat{\boldsymbol{\mu}} | \Phi_{K} \rangle.$

Dipoles of the two A" states which cross, abruptly switch character (and sign) making for a convenient properties based diabatization.

Samanta K, Beames JM, Lester MI, Subotnik JE. JCP 2014;141(13):134303.

AISSOURI

Fitted Surfaces: Energies

~1500 symmetry unique data fit with committee of neural networks Energies at MRCI-F12(Q):GDW(18)-CASSCF(13e,10o)/VTZ-F12 level



Fitted Surfaces: Energies





Fitted Surfaces: Transition Dipoles



Ground State Vibrational Levels



- Vibrational states computed using large DVR basis with numerically exact kinetic operator
- MCTDH method yields same energies
- Initial test of lowest 76 levels finds RMSE of 14.1 cm⁻¹ confirming accuracy of PES
- Error much larger without Davidson correction (~100 cm⁻¹)
- Error doesn't increase at higher energies??...



Ground State Vibrational Levels

	Modal	Calc	Calc-ZPE	EXPT*	Expt-Calc
1	0,0,0	1870.239	0.000	0.00	0.00
2	0,1,0	2611.628	741.389	749.65	8.26
3	1,0,0	3195.168	1324.929	1319.79	-5.14
4	0,2,0	3351.313	1481.074	1498.34	17.27
5	0,0,1	3484.052	1613.813	1616.85	3.04
6	1,1,0	3932.341	2062.102	2063.12	1.02
7	0,3,0	4091.960	2221.721	2246.04	24.32
8	0,1,1	4214.738	2344.499	2355.15	10.65
9	2,0,0	4507.912	2637.673	2627.34	-10.33
10	1,2,0	4668.544	2798.305	2805.60	7.30
11	1,0,1	4775.601	2905.362	2906.07	0.71
12	0,4,0	4834.125	2963.886	2993.00	29.11
13	0,2,1	4943.023	3072.784	3092.48	19.70
14	0,0,2	5069.031	3198.792	3201.44	2.65
15	2,1,0	5239.784	3369.545	3364.57	-4.97
16	1,3,0	5405.001	3534.762	3547.10	12.34
17	1,1,1	5502.523	3632.284	3637.84	5.56
18	0,5,0	5577.652	3707.413	3738.60	31.19
19	0,3,1	5672.651	3802.412	3829.34	26.93
20	0,1,2	5787.813	3917.574	3929.12	11.55
21	3,0,0	5808.052	3937.813	3922.61	-15.20
22	2,2,0	5971.248	4101.009	4100.58	-0.43
23	2,0,1	6052.372	4182.133	4179.94	-2.19
24	1,4,0	6141.782	4271.543	4286.82	15.28
25	1,2,1	6228.114	4357.875	4369.10	11.23
26	0,6,0	6321.974	4451.735	4482.57	30.84
27	1,0,2	6329.630	4459.391	4461.07	1.68
28	0,4,1	6404.246	4534.007	4564.22	30.21
29	0,2,2	6505.384	4635.145	4656.34	21.19
30	3,1,0	6533.822	4663.583	4652.00	-11.58

RMSE for lowest 76 levels = 14.1 cm^{-1}

RMSE for lowest 125 levels = 16.5 cm⁻¹

		Bend				
			E(n)-E(n-1)		E(n)-E(n-1)	
		calc		expt		Expt-calc
1	0,0,0	0.000		0.00		0.00
2	0,1,0	741.389	741.389	749.65	749.65	8.26
4	0,2,0	1481.074	739.685	1498.34	748.69	17.27
7	0,3,0	2221.721	740.647	2246.04	747.70	24.32
12	0,4,0	2963.886	742.165	2993.00	746.96	29.11
18	0,5,0	3707.413	743.527	3738.60	745.60	31.19
26	0,6,0	4451.735	744.322	4482.57	743.97	30.84
37	0,7,0	5195.909	744.174	5224.55	741.98	28.64
50	0,8,0	5938.550	742.641	5965.61	741.06	27.06
67	0,9,0	6677.777	739.227	6705.23	739.62	27.45
88	0,10,0	7411.096	733.319	7443.09	737.86	31.99

			Sym stretch						
1	L	0,0,0	0.000		0.00		0.00		
3	3	1,0,0	1324.929	1324.929	1319.79	1319.79	-5.14	calc anharm	expt anharm
9)	2,0,0	2637.673	1312.744	2627.34	1307.55	-10.33	-12.185	-12.24
2	1	3,0,0	3937.813	1300.140	3922.61	1295.27	-15.20	-12.604	-12.28
3	8	4,0,0	5225.218	1287.405	5205.81	1283.20	-19.41	-12.735	-12.07
6	3	5,0,0	6499.670	1274.452	6475.05	1269.24	-24.62	-12.953	-13.96
9	8	6,0,0	7760.091	1260.421	7730.08	1255.03	-30.01	-14.031	-14.21

		asym stretch						
1	0,0,0	0.000		0.00		0.00		
5	0,0,1	1613.813	1613.813	1616.85	1616.85	3.04	calc anharm	expt anharm
14	0,0,2	3198.792	1584.979	3201.44	1584.59	2.65	-28.834	-32.26
31	0,0,3	4753.133	1554.341	4754.21	1552.77	1.08	-30.638	-31.82
58	0,0,4	6276.629	1523.496	6275.98	1521.77	-0.65	-30.845	-31.00
99	0,0,5	7767.589	1490.960	7766.28	1490.30	-1.31	-32.536	-31.47

* Delon, A. and Jost, R., 1991. The Journal of chemical physics, 95(8), pp.5686-5700.

UV absorption spectrum: The reflection principle and T-dep



Figure 5. The evolution of the gross (envelope) cross section from 200 to 600 K. It should be noted that the temperature effect increases dramatically to the IR. The temperature dependence on the blue side, below 350 nm, is more difficult to characterize, mainly because N₂O₄ absorbs in that range. Then, it is difficult to discriminate between the temperature effect on the NO₂ absorption cross section itself and on the NO₂/N₂O₄ ratio effect.

Rémy Jost studied NO₂ extensively for many years and provided a clear foundation of understanding

Calculated UV Spectrum (zero Kelvin) MCTDH 100-500 fs propagations



- Total integrated cross section (15,000 40,000) matches experiment to within 1.5%
- At 294 K, only 2.5% of population is in higher vibrational levels, but many hundreds of rotational states are populated





Cross Section (cm^2)



Calculation at 70 K reproduces some of the detailed structure in low-energy wing coming from both $X \rightarrow A$ and $X \rightarrow B/C$ contributions, but doesn't match expt. resolution.





Temperature-weighted NO₂ UV absorption spectrum 5x10⁻¹⁹ Experiment (Davidson 233K) _____ 4.5x10⁻¹⁹ X --> A,B,C. Jm=20 (V=0). T=70K _____ X --> A. Jm=20 (V=0). T=70K _____ 4x10⁻¹⁹ X --> B,C. Jm=20 (V=0). T=70K _____ 3.5x10⁻¹⁹ X --> A,B,C. Jm=0 3x10⁻¹⁹ 2.5x10⁻¹⁹ 2x10⁻¹⁹ 1.5x10⁻¹⁹ 1x10⁻¹⁹ 5x10⁻²⁰ 0 15000 17000 18000 19000 20000 21000 22000 16000

Cross Section (cm²)

Energy (cm⁻¹)



Simulations of higher temperature



7

MISSOURI

Comparison to Paulsen and Huffman (669 K)



D. E. Paulsen and R. E. Huffman Air Force Geophysics Laboratory Hanscom AFB (1976) Ground vibrational state remains dominant, while a rapidly increasing number of states make small contributions.



Temperature scans using the tool



Temperature-weighted NO₂ UV absorption spectrum

125 contributing vibrational states



Mode specific effects and contributions



More diffuse initial state spreads wings of spectrum.
Peak structure shifted by mode energy.

Result is B/C transition clearly projects prob. dens. onto energy axis.







Mode specific effects and contributions Bend progression

(0,1,0)

(0,0,0)



cm⁴)

5







(0,3,0)





(0,4,0)







Mode specific effects and contributions

(1,0,0)

(0,0,0)



(cm∠)

5



(2,0,0)



Cross Section (cm²)

(0,0,1)



(0,0,2)



(0,1,2)



T)C

Mode specific effects and contributions Interpretation based on energy and property surfaces





Conclusions and Future Work

- Fitted energy and property surfaces combined with an accurate quantum dynamics approach such as MCTDH can accurately predict spectra even in cases with complex electronic structure and dynamics
- Even simpler surveys using the reflection principle can benefit substantially from including the transition dipole
- The coupled energy and properties surfaces generated for this study provide the foundation for a variety of dynamical investigations of this remarkably complex system



Acknowledgements

\circ Collaborators

- Dynamics Hua Guo (New Mexico), Fabien Gatti (Paris-Saclay), François Lique (Le Havre)
- o Ro-vibrational Calculations Xiao-Gang Wang and Tucker Carrington Jr. (Queen's University)
- Experimental Spectroscopy Scott A. Reid (Marquette), Michael Heaven (Emory), Michael McCarthy (Harvard-Smithsonian), David Osborn (Sandia)
- Missouri S&T Graduate Students and Postdocs
 Bradley Welch, Steve Ndengué, Sangeeta Sur, Richard Dawes, Andrew Powell, Ernesto Quintas Sánchez http://web.mst.edu/~dawesr/people.html





